Antimicrobial and Cytotoxic Phenolic Glycoside Esters from the New Zealand Tree *Toronia toru*

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Two new compounds, 4-hydroxyphenyl 6-*O*-(4-hydroxy-2-methylenebutanoyl)- β -D-glucopyranoside (**1**) and 4-hydroxyphenyl 6-*O*-[(3*R*)-3,4-dihydroxy-2-methylenebutanoyl]- β -D-glucopyranoside (**2**), have been isolated from foliage of *Toronia toru*. Compound **2** was the main antimicrobial component (7 mg/g of dry foliage) of the crude extract and also showed significant cytotoxic activity against P-388 leukemia cells.

We have been screening New Zealand plants for bioactive natural products with the aim of developing new pharmaceuticals and agrochemicals.¹ An extract from *Toronia toru* (Cunn.) L. Johnson et B. Briggs (Maori name "toru") showed strong cytotoxic and antimicrobial activity. Similar bioactivity was noted in an earlier screening program,² but no Maori traditional medicinal uses have been recorded for this plant.^{3,4} *T. toru* is a small tree native to the north of the North Island of New Zealand.⁵ It was previously known as *Persoonia toru* A. Cunn.⁶ but was recently assigned as the only species in the genus *Toronia*, of the family Proteaceae.⁷

We could not find any references to previous chemical studies on *T. toru*, but in view of the compounds identified in the present investigation, it is noteworthy that arbutin (**3**) and hydroquinone have been detected in several species of Proteaceae.⁸ We now report the isolation and structure identification of two new phenolic glycoside esters **1** and **2**, which are the main antimicrobial and cytotoxic components from *T. toru*. During the course of this work, compound **2** was independently found to be the main antimicrobial component in the fruits of an Australian tree *Persoonia linearis* × *pinifolia*.⁹

A crude extract of *T. toru* leaf and stem showed broadspectrum antibacterial activity against both Grampositive bacteria (*Bacillus subtilis, Dermatophilus congolensis*, and *Staphylococcus aureus*) and Gram-negative bacteria (*Escherichia coli* and *Pseudomonas aeruginosa*), and against the dermatophyte *Trichophyton mentagrophytes.* The *T. toru* extract also showed cytotoxic activity against both BSC monkey kidney cells and P-388 leukemia cells (P-388; IC₅₀ ca. 8 μ g/mL).

Assays against *B. subtilis* were used to guide the isolation of the main antibacterial component(s). Reversed-phase (C_{18}) flash chromatography gave most of the antibacterial activity in the highest polarity fractions, eluted with H_2O . Further reversed-phase chromatography was carried out on an ethyl-functionalized stationary phase, as this was expected to be more retentive of polar compounds. Two related compounds, **1** and **2**, were obtained pure from this column. Compound **2** was the main antibacterial component in the extract.

The NMR spectra of compounds **1** and **2** (Table 1) both showed ¹H signals at 4.96 ppm (d, 7 Hz) and ¹³C signals at 103.6 and 103.7 ppm (d, ¹ $J_{C-H} = 164$ Hz), suggestive of β -glucopyranosides. These spectra also contained signals, at 6.98 and 6.99 ppm (2H, d, 9 Hz) and 6.82 and 6.84 ppm (2H, d, 9 Hz), as found in hydroquinone derivatives. The UV spectra of compounds **1** and **2** were similar to the UV spectrum of pyroside (**4**). Therefore, it seemed possible that these compounds from *T. toru* were derivatives of arbutin (**3**). This hypothesis was supported by close matches in the ¹H- and ¹³C-NMR data of **1** and **2** with those of pyroside (**4**).¹⁰



The ¹H- and ¹³C-NMR spectra of compound **1** showed the following groups additional to the arbutin skeleton: ester C=O, C=CH₂, C-CH₂-C, and C-CH₂-O (Table 1). These required a molecular formula of $C_{17}H_{22}O_9$, which was supported by HRFABMS. The full structure of 1 was shown by a two- and three-bond ¹H-¹³C correlation NMR experiment (HMBC, Table 1). This showed the presence of a 2-methylene-4-hydroxybutanoic acid unit, as well as the expected β -glucopyranoside and hydroquinone units. The linkages between these units were established by correlations from both glucose H-6's to the ester carbonyl C-1', and from the glucose H-1 to one aromatic signal, C-1". Therefore, this compound has the previously unreported structure, 1. A few other glycosides esterified at C-6 with 2-methylene-4-hydroxybutanoic acid are known,^{11,12} with the simplest analogue of **1** being 6-tuliposide A (**5**).¹² The

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Table 1.	¹ H- and	¹³ C-NMR	Data d	of Com	pounds	1 an	d 2 ^a
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		1			
position	$^{1}\mathrm{H}^{b}$	¹³ C	HMBC, ¹ H to ¹³ C correlations	¹ H	¹³ C
1	4.96 (d, 7)	103.6 (br d, 164)	152.8	4.96 (d, 7)	103.7
2	3.59 (dd, 7, 9)	75.5 (dd, 147, 6)	103.6, 78.2	3.5 - 3.7	75.6
3	3.63 (t, 9)	78.2 (br d, 148)	75.5	3.5 - 3.7	78.1
4	3.53 (t, 9)	72.7 (br d, 144)	78.2, 66.4	3.5 - 3.7	72.6
5	3.80 (ddd, 2, 7, 9)	76.2 (br d, 148)	72.7	3.82 (ddd, 2, 7, 9)	76.2
6	4.56 (dd, 2, 12)	66.4 (t, 151)	171.0, 72.7	4.54 (dd, 2, 12)	66.4
6	4.34 (dd, 7, 12)		171.0, 76.2	4.40 (dd, 7, 12)	
1'		171.0 (s)			169.5
2'		138.8 (s)			141.5
3′	2.52 (t, 7)	36.7 (br t, 131)	171.0, 138.8, 131.6, 62.6	4.61 (dd, 6, 3)	73.0
4'	3.67 (t, 7)	62.6 (tt, 144, 5)	138.8, 36.7	3.5-3.7	67.0
5'	5.78 (br s)	131.6 (tt, 161, 6)	171.0, 36.7	6.04 (br s)	130.9
5'	6.27 (s)		171.0, 138.8, 36.6	6.42 (s)	
1″		152.8 (s)			152.8
2″	6.98 (d, 9)	121.0 (dd, 161, 5)	153.9, 152.8, 121.0, 118.8	6.99 (d, 9)	121.1
3″	6.82 (d,9)	118.8 (dd,161,5)	152.8,118.8	6.84 (d,9)	118.8
4″		153.9 (s)			153.9
5″	6.82 (d,9)	118.8 (dd,161,5)	152.8,118.8	6.84 (d,9)	118.8
6″	6.98 (d,9)	121.0 (dd,161,5)	153.9,152.8,121.0,118.8	6.99 (d,9)	121.1

^{*a*} In D₂O, δ in ppm (multiplicity, *J* in Hz). ^{*b*} One bond ¹H–¹³C correlations confirmed by HMQC.

¹H- and ¹³C-NMR data reported for the β -anomer of **5** match our data for the ester unit of **1** very well.¹³



The HRFABMS of compound **2** showed the presence of one further oxygen atom compared to 1. The ¹H- and ¹³C-NMR spectra of **2** showed very similar signals to the hydroquinone and β -glucopyranoside signals of **1** (Table 1). The NMR spectra of **2** did not show the CH₂ -3' signals of 1, but did show an additional -CHOgroup [¹H 4.61 ppm (dd, 6, 3 Hz); ¹³C 73.0 ppm]. Therefore, this compound was the 3,4-dihydroxy 2-methylbutanoate ester 2. 1-Tuliposide B (6) has the same ester group,¹² but structure **2** has not previously been reported in the literature. MacLeod et al. have also isolated compound 2 from a hybrid between P. pinifolia and P. linearis.9 They determined the relative stereochemistry of 2 by X-ray crystallography, and the absolute stereochemistry of CHOH-3' by hydrolysis of 2 to lactone 7.9 Compound 2 from T. toru had the same ¹Hand ¹³C-NMR spectra as the compound from *P. linearis* × pinifolia and a similar optical rotation.

Compounds 1 and 2 were major components of our bulk extract of T. toru foliage, with isolated yields of 7 mg/g dry foliage for 2 and 3 mg/g for 1. Characteristic signals of 1 and 2 were prominent in the ¹H-NMR spectrum of the crude extract. We used biological assays and ¹H-NMR spectra to check whether 1 and 2 were present in other collections of T. toru. Five different T. toru samples all showed the same strong antimicrobial and cytotoxic activity as our original collection, and all showed characteristic ¹H-NMR peaks of 1 and 2. Therefore, these results suggest that these novel phenolic glycoside esters are characteristic of *T*. *toru* foliage. It would be interesting to know whether they also occur widely in the closely related genus Persoonia. MacLeod et al. only reported 2 in their extract of *P. linearis* \times *pinifolia* fruits.⁹

The antimicrobial and cytotoxic activities of new compounds 1 and 2 are summarized in Table 2. Compound **2** showed broad-spectrum antibacterial activity, plus antifungal activity against T. mentagrophytes, and cytotoxic activity against both P-388 leukemia and BSC monkey kidney cells. Compound 1 was only active (at the doses tested) against the BSC cells. The antimicrobial and cytotoxic activity of compound 2 could be due to formation of lactone 7, which is both antimicrobial and cytotoxic.^{12,14} α -Methylene lactone 7 is probably toxic because of Michael addition to biological nucleophiles. The corresponding lactone 8, which might be released by hydrolysis of 1, was antimicrobial and cytotoxic in our assays (Table 2). The differing biological activities of 1 and 2 could be due to different rates of lactone formation in the assay media or within the target cells. Hydroquinone, another possible product from hydrolysis of 1 and 2, was also antimicrobial and cytotoxic (Table 2), although arbutin (3) was inactive. These phenolic glycoside esters (1 and 2) may represent another class of "post-inhibitin" in plants: inactive glycosides from which active toxins are released by enzymic hydrolysis, following microbial invasion or herbivore attack on foliage.¹⁵



Experimental Section

General Experimental Procedures. All solvents were distilled before use and were removed by rotary evaporation at temperatures up to 45 °C. Octadecylfunctionalized Si gel (Aldrich) was used for C_{18} chromatography, and ethyl-functionalized Si gel (International Sorbent Technology) was used for C_2 chromatography. TLC was carried out using Merck DC-Plastikfolien Kieselgel 60 F₂₅₄. MS, IR, and UV spectra were recorded on Kratos MS80, Perkin-Elmer 1600, and JASCO 7800 UV-vis spectrometers, respectively. Optical rotations were recorded on a JASCO DIP-1000 polarimeter. NMR spectra, in D₂O solution at 25 °C,

Table 2.	Biological Assay	Results for	Compounds 1	l and 2 and	Related	Compounds
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	antimicrobial activity ^a					cytotoxicity	
compound	dose (µg/disk)	B. subtilis	E. coli	P. aeruginosa	T. mentagrophytes	BSC ^b	P-388 ^c
1	120	0	0	0	0	100	>25
	60	0	0	0	0	100	
	30	0	0	0	0	100	
	15	NT	NT	NT	NT	50	
2	120	5	4	3	5	100	3
	60	1	2	1	0	100	
	30	0	0	0	0	50	
3	120	0	0	0	0	50	>25
hydroquinone	120	9	5	3	10	100	3
	60	4	0	3	0	100	
	30	3	0	2	0	75	
tulipalin A (8)	120	10	5	4	8	100	0.8
	60	0	0	0	0	100	
	30	0	0	0	0	100	
	15	0	0	0	0	100	

^{*a*} Width of zone of growth inhibition (mm), NT = not tested. ^{*b*} % of well showing cytotoxic effects. ^{*c*} IC₅₀ (μ g/mL).

were recorded at 300 MHz for ¹H and 75 MHz for ¹³C on a Varian VXR-300 spectrometer. Chemical shifts were referenced to the main signal of 3-(trimethylsilyl)-1-propanesulfonic acid, sodium salt (DSS, Aldrich) at 0 ppm. Arbutin (Sigma), lactone 8 (Aldrich), and hydroquinone were commercial samples.

Collection and Screening. Foliage (leaves and stems) was collected from a tree in the Dunedin Botanical Gardens in September 1995. A voucher specimen, 950919-01, has been kept in the Plant Extracts Research Unit Collection. Further samples of T. toru foliage were obtained in August 1996, from the Auckland Regional Botanic Gardens (voucher 960807-01) and from Otari Native Botanical Gardens, Wellington (plants from four different source locations, vouchers 960820-01 to -04). Foliage was air-dried at 35 °C for 3 days. Initial screening was carried out using an extract produced by shaking milled material (5.0 g) overnight in EtOH (95% EtOH-5% H₂O; 50 mL). A subsample (1 mL) of this extract was dried, dissolved in D_2O (1 mL), and filtered prior to ¹H-NMR analysis (200 MHz, Varian Gemini). Spectra, referenced to HOD at 4.70 ppm, were examined for characteristic peaks of 1 at 6.15 ppm (s), 5.65 ppm (s), and 2.30 ppm (t), and of 2 at 6.30 ppm (s) and 5.90 ppm (s).

Extraction and Isolation. A bulk extract of dried plant material (950919-01, 103 g) was prepared by blending with EtOH (1 \times 600 mL and 2 \times 450 mL). The solvent was removed from the combined, filtered extracts to give a green gum (20.9 g, inhibition zone against *B. subtilis* 2 mm, 120 μ g/disk; abbreviated as Bs 2 @120). A subsample of this extract (6 g) was precoated onto 10 g of C_{18} Si gel and loaded onto a 50-g C_{18} Si gel column, preconditioned with H_2O . The column was developed in 37-mL steps with H₂O through to MeCN, finishing with 1:1 MeCN-CHCl₃. The most active fractions were eluted with H₂O (1.62 g, Bs 2 @ $120 \,\mu g/disk$). A subsample (1 g) of these active fractions was precoated onto 2 g of C2 Si gel and loaded onto a 20-g C₂ Si gel column, preconditioned with H₂O. This column was developed in 20-mL steps with H₂O and combined on the basis of TLC on Si gel. The most potent antimicrobial component, from combined fractions 4-8 (Bs 6 @ 120 µg/disk) was pure 2 (130 mg). A later sample, from combined fractions 13-15 (Bs 0 @ $120 \,\mu g/$ disk) contained pure 1 (62 mg).

4-Hydroxyphenyl 6-O-(4-hydroxy-2-methylene**butanoyl**)-β-**D**-glucopyranoside (1): pale orange gum; $[\alpha]^{19}_{D}$ +18° (*c* 2.5, MeOH); UV (EtOH) λ_{max} (log ϵ) 285 (3.61) nm; IR (Nujol) v_{max} 3274 (br), 1708, 1634, 1511 cm⁻¹; ¹H- and ¹³C-NMR data in Table 1; FABMS (glycerol + NaCl) m/z 393.1149 (M + Na⁺, C₁₇H₂₂O₉-Na requires 393.1161); Si gel TLC (100:13.5:10 EtOAc-MeOH-H₂O) R_f 0.3, UV-active plus blue with Berlin Blue reagent.

4-Hydroxyphenyl 6-O-[(3R)-3,4-dihydroxy-2-meth**ylenebutanoyl**]-β-D-glucopyranoside (2): colorless gum; $[\alpha]^{19}$ – 48° (c 2.5, MeOH); UV (EtOH) λ_{max} (log ϵ) 285 (3.89) nm; IR (Nujol) v_{max} 3305 (br), 1710 (br), 1634 (br) cm⁻¹; ¹H- and ¹³C-NMR data in Table 1, ¹H-NMR (in CD₃COCD₃) and ¹³C-NMR (in CD₃OD) spectra match MacLeod et al.;9 FABMS (glycerol) m/z 387.1283 (MH+ C₁₇H₂₃O₁₀ requires 387.1291); Si gel TLC (100:13.5:10 EtOAc-MeOH-H₂O) R_f 0.3, UV-active plus blue with Berlin Blue reagent.

Biological Testing. Details of antimicrobial and cytotoxicity assays have been given previously.¹⁶

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